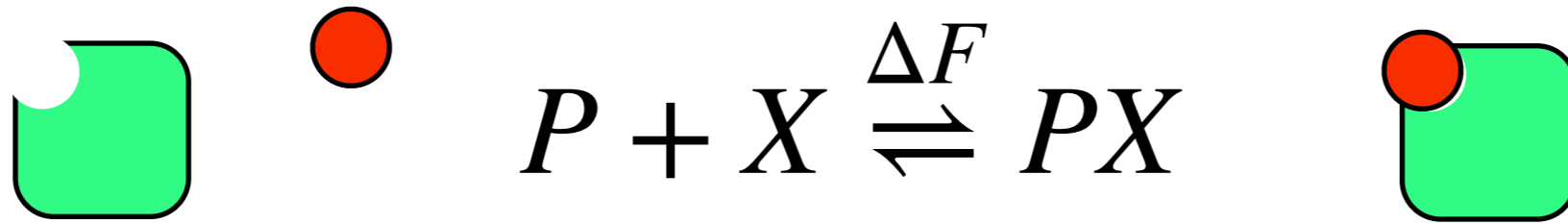


# Free energy calculations with *alchemlyb*

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# Binding free energy



- measure of how strong a protein  $P$  and a “ligand”  $X$  stick together
- key quantity in quantitative mechanistic explanations of biological processes and in drug discovery
- **free energy** (i.e., averaged over all other degrees of freedom (such as solvent, protein motions, ...))

$$\exp[-A(T, V, N)/kT] = \int dp^{3N} dx^{3N} e^{-H(p, x)/kT} \quad H(p, x) = \sum_{i=1}^N \frac{\mathbf{p}_i^2}{2m_i} + U(x)$$

- **free energy difference**

$$\Delta A = A_{\text{bound}} - A_{\text{unbound}}$$

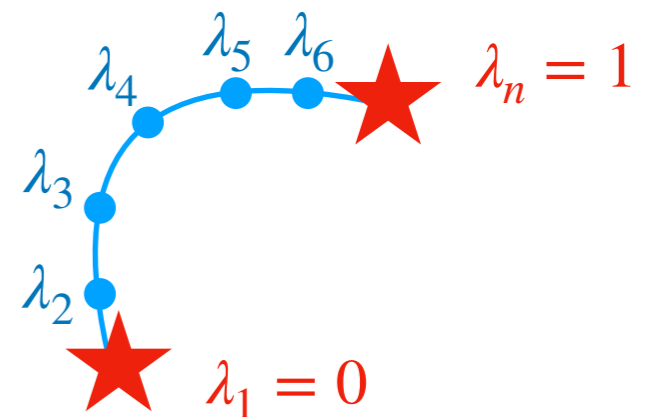
$$\Delta A = -\ln \frac{\int dx^{3N} \exp[-U(x)/kT] \chi_{\text{bound}}(x)}{\int dx^{3N} \exp[-U(x)/kT] \chi_{\text{unbound}}(x)}$$

# Alchemical free energy calculations

- force fields for  $H$ , molecular dynamics (MD) simulations for sampling
- free energy is a **state function**: generate **non-physical paths** between **physical end states** (bound/unbound)
- use **stratification** (“windows”) of path with parameter  $\lambda$

$$H(\lambda) = (1 - \lambda)H_{\text{bound}} + \lambda H_{\text{unbound}}, \quad 0 \leq \lambda \leq 1$$

$$U_{\text{Coulomb}}(\mathbf{x}_1, \mathbf{x}_2; \lambda) = \frac{1}{4\pi\epsilon_0} \frac{(1 - \lambda)q_1q_2}{|\mathbf{x}_1 - \mathbf{x}_2|}$$



# Methods

- “Free Energy Perturbation” (FEP): Zwanzig FEP, BAR, MBAR (overlaps of distributions)

$$\Delta U(x) = U_{\lambda_{n+1}}(x) - U_{\lambda_n}(x)$$

$$\Delta A = -kT \ln \langle \exp[-\Delta U(x)/kT] \rangle_1, \quad \text{with } \Delta U(x) = U_1(x) - U_0(x) \quad \text{FEP}$$

$$\exp(-\Delta A/kT) = \frac{\langle 1 + \exp[(\Delta U - C)/kT]^{-1} \rangle_0}{\langle 1 + \exp[(\Delta U - C)/kT]^{-1} \rangle_1} \exp(-C/kT) \quad \text{BAR}$$

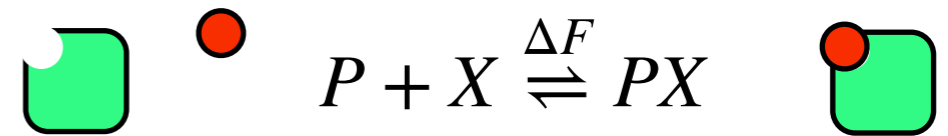
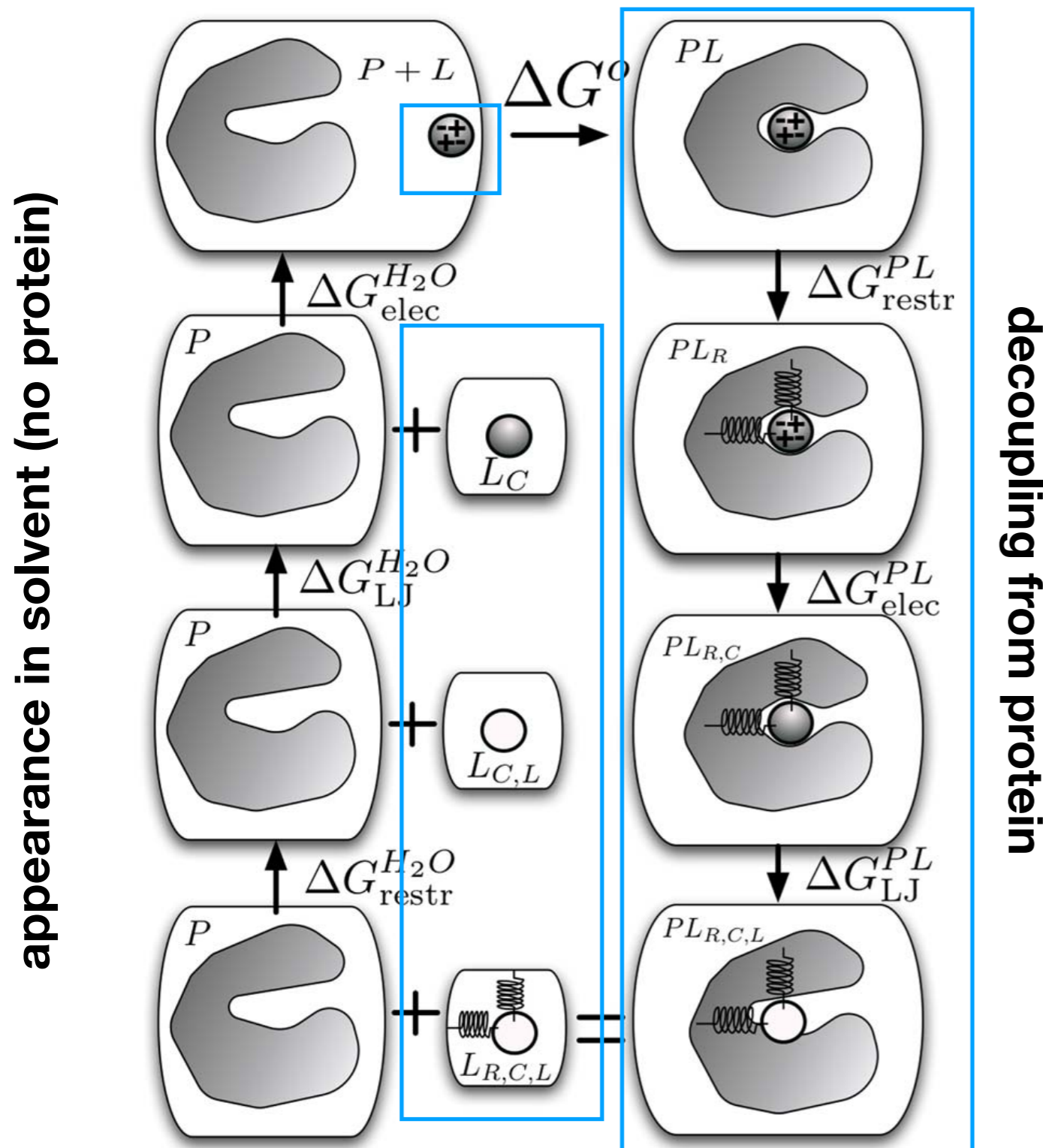
$$\Delta A/kT = C/kT - \ln \frac{n_1}{n_2}$$

(MBAR is more complicated: uses overlaps between all windows)

- Thermodynamic Integration (TI): TI

$$\Delta A = \int_0^1 d\lambda \left\langle \frac{\partial H(\lambda)}{\partial \lambda} \right\rangle_\lambda \quad \text{TI}$$

# Thermodynamic cycle for absolute binding free energy calculations



**MD simulations with multiple  $\lambda$**

- 5 – 50  $\lambda$ -windows per free energy component
- 10 ns – 250 ns per window

D. L. Mobley, J. D. Chodera, and K. A. Dill. On the use of orientational restraints and symmetry corrections in alchemical free energy calculations. 125:084902, 2006.

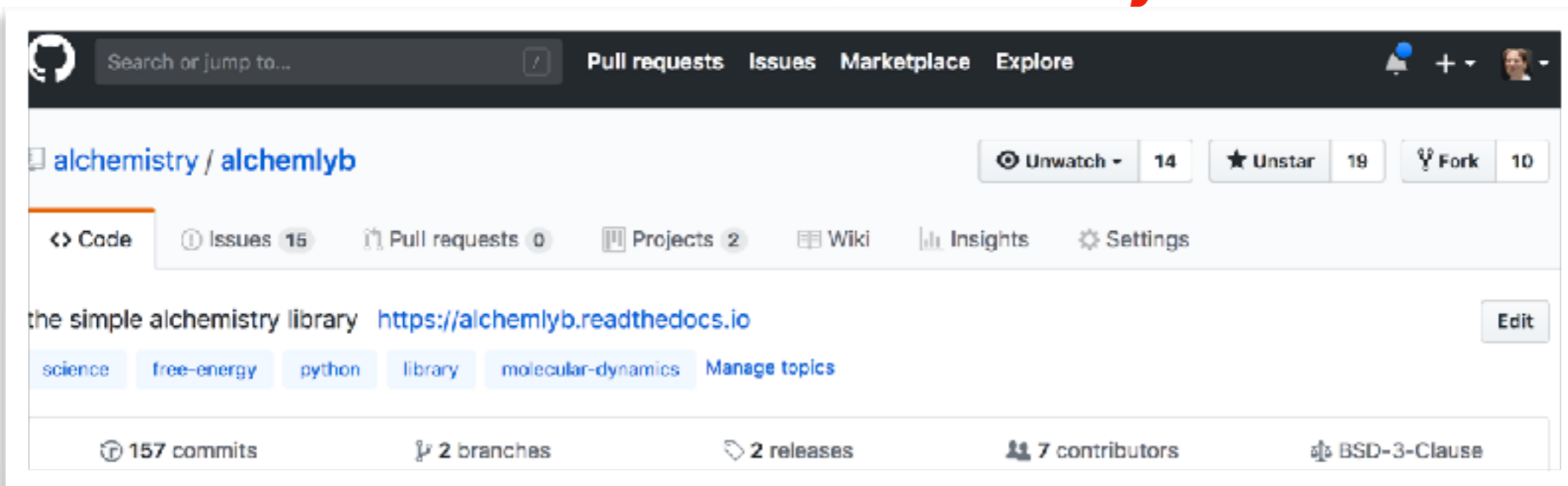
# Simulation output

Per timestep

- FEP:  $\Delta U_{\lambda_i, \lambda_j} \forall j$  for window  $i$
- TI:  $\frac{\partial U(x; \lambda)}{\partial \lambda} \Big|_{\lambda_i}$

But: different file formats for different codes (Gromacs, Amber, NAMD, ...)

solution: common interface via ***alchemlyb***



# alchemyb

- O Beckstein (ASU), D Mobley (UC Irvine), M Shirts (U Colorado, Boulder)
- **David Dotson** (ASU), Dominik Wille (Freie Univ. Berlin)
- Silicon Therapeutics (STX) (Bryce Allen, Shuai Liu)

The screenshot shows the GitHub repository page for 'alchemyb: the simple alchemy library'. The repository name is prominently displayed at the top. Below the name, there are several status indicators: 'DOI 10.5281/zenodo.583647', 'docs passing', 'build passing', and 'codecov 98%'. The 'build passing' and 'codecov 98%' indicators are highlighted with a red dashed box. Below these indicators, there is a link to the documentation: 'the simple alchemy library https://alchemyb.readthedocs.io'. There are also topic tags: 'science', 'free-energy', 'python', 'library', and 'molecular-dynamics'. At the bottom of the repository page, there are statistics: '157 commits', '2 branches', '2 releases', and '7 contributors'. The license 'BSD-3-Clause' is also displayed and highlighted with a red dashed box.

alchemyb: the simple alchemy library

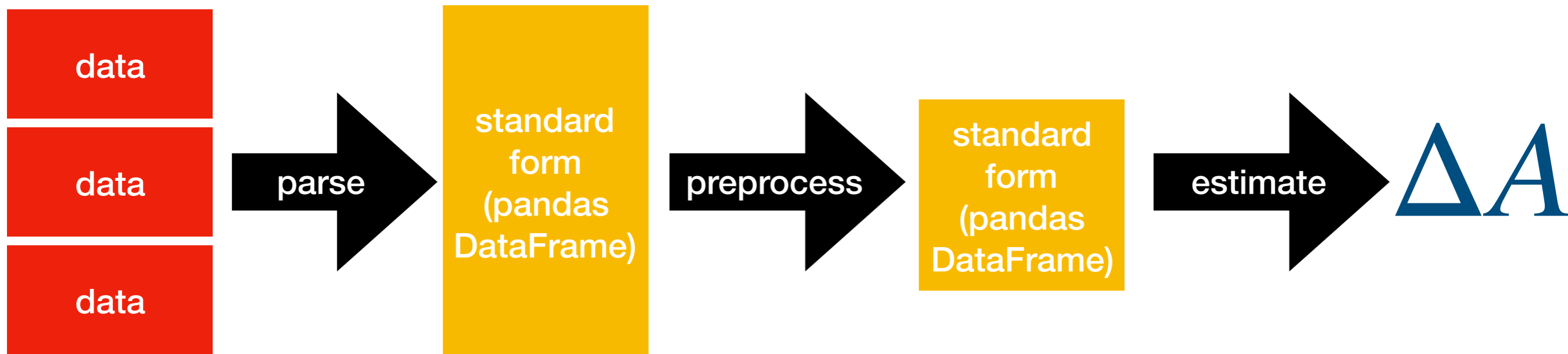
DOI 10.5281/zenodo.583647 docs passing build passing codecov 98%

the simple alchemy library <https://alchemyb.readthedocs.io>

science free-energy python library molecular-dynamics Manage topics

157 commits 2 branches 2 releases 7 contributors BSD-3-Clause

# alchemlyb: basic idea



## preprocessors

- slicing
- statistical inefficiency
- equilibrium detection
- ...

## estimators

- TI
- BAR
- MBAR
- ...

$$\Delta U_{\lambda_i, \lambda_j} \forall j$$

$$\left. \frac{\partial U(x; \lambda)}{\partial \lambda} \right|_{\lambda_i}$$



As a usage example, we'll use `TI` to calculate the free energy of solvation of benzene in water. We'll use the benzene-in-water dataset from `alchemtest.gmx`:

```
>>> from alchemtest.gmx import load_benzene
>>> bz = load_benzene().data
```

and parse the datafiles separately for each alchemical leg using `alchemlyb.parsing.gmx.extract_dHdl()` to obtain `dHdl` gradients:

```
>>> from alchemlyb.parsing.gmx import extract_dHdl
>>> import pandas as pd

>>> dHdl_coul = pd.concat([extract_dHdl(xvg, T=300) for xvg in bz['Coulomb']])
>>> dHdl_vdw = pd.concat([extract_dHdl(xvg, T=300) for xvg in bz['VDW']])
```

We can now use the `TI` estimator to obtain the free energy differences between each  $\lambda$  window sampled. The `fit()` method is used to perform the free energy estimate, given the gradient data:

```
>>> from alchemlyb.estimators import TI

>>> ti_coul = TI()
>>> ti_coul.fit(dHdl_coul)
TI(verbose=False)

# we could also just call the `fit` method
# directly, since it returns the `TI` object
>>> ti_vdw = TI().fit(dHdl_vdw)
```

The sum of the endpoint free energy differences will be the free energy of solvation for benzene in water. The free energy differences (in units of  $k_B T$ ) between each  $\lambda$  window can be accessed via the `delta_f_` attribute:

```
>>> ti_coul.delta_f_
      0.00      0.25      0.50      0.75      1.00
0.00  0.000000  1.620328  2.573337  3.022170  3.089027
0.25 -1.620328  0.000000  0.953009  1.401842  1.468699
0.50 -2.573337 -0.953009  0.000000  0.448832  0.515690
0.75 -3.022170 -1.401842 -0.448832  0.000000  0.066857
1.00 -3.089027 -1.468699 -0.515690 -0.066857  0.000000
```

So we can get the endpoint differences (free energy difference between  $\lambda = 0$  and  $\lambda = 1$ ) of each with:

```
>>> ti_coul.delta_f_.loc[0.00, 1.00]
3.0890270218676896

>>> ti_vdw.delta_f_.loc[0.00, 1.00]
-3.0558175199846058
```

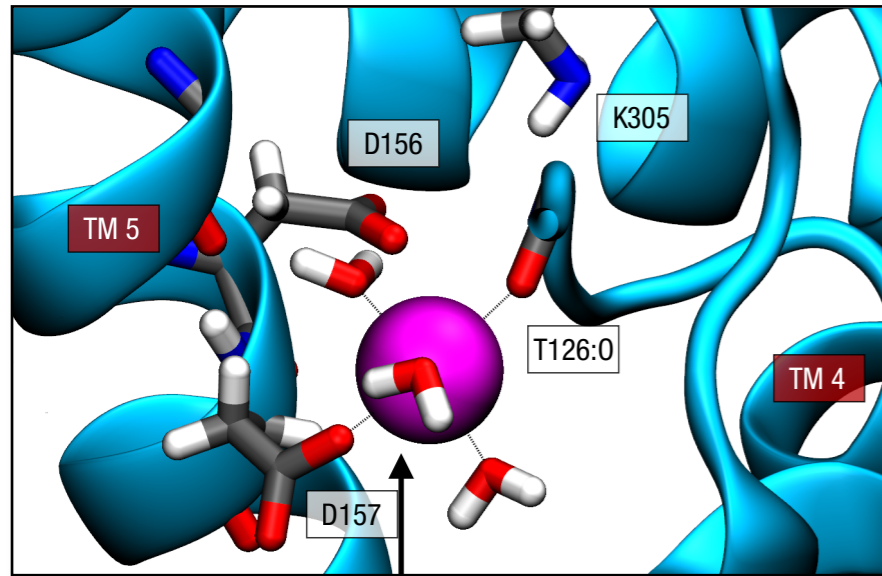
giving us a solvation free energy in units of  $k_B T$  for benzene of:

```
>>> ti_coul.delta_f_.loc[0.00, 1.00] + ti_vdw.delta_f_.loc[0.00, 1.00]
0.033209501883083803
```

In addition to the free energy differences, we also have access to the errors on these differences via the `d_delta_f_` attribute:

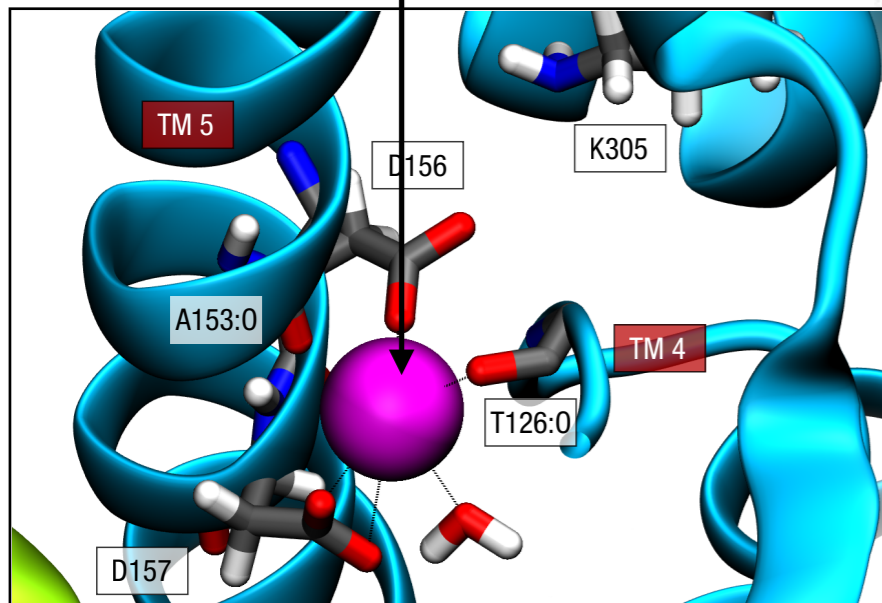
```
>>> ti_coul.d_delta_f_
      0.00    0.25    0.50    0.75    1.00
0.00  0.000000  0.009706  0.013058  0.015038  0.016362
0.25  0.009706  0.000000  0.008736  0.011486  0.013172
0.50  0.013058  0.008736  0.000000  0.007458  0.009858
0.75  0.015038  0.011486  0.007458  0.000000  0.006447
1.00  0.016362  0.013172  0.009858  0.006447  0.000000
```

# NapA elevator mechanism

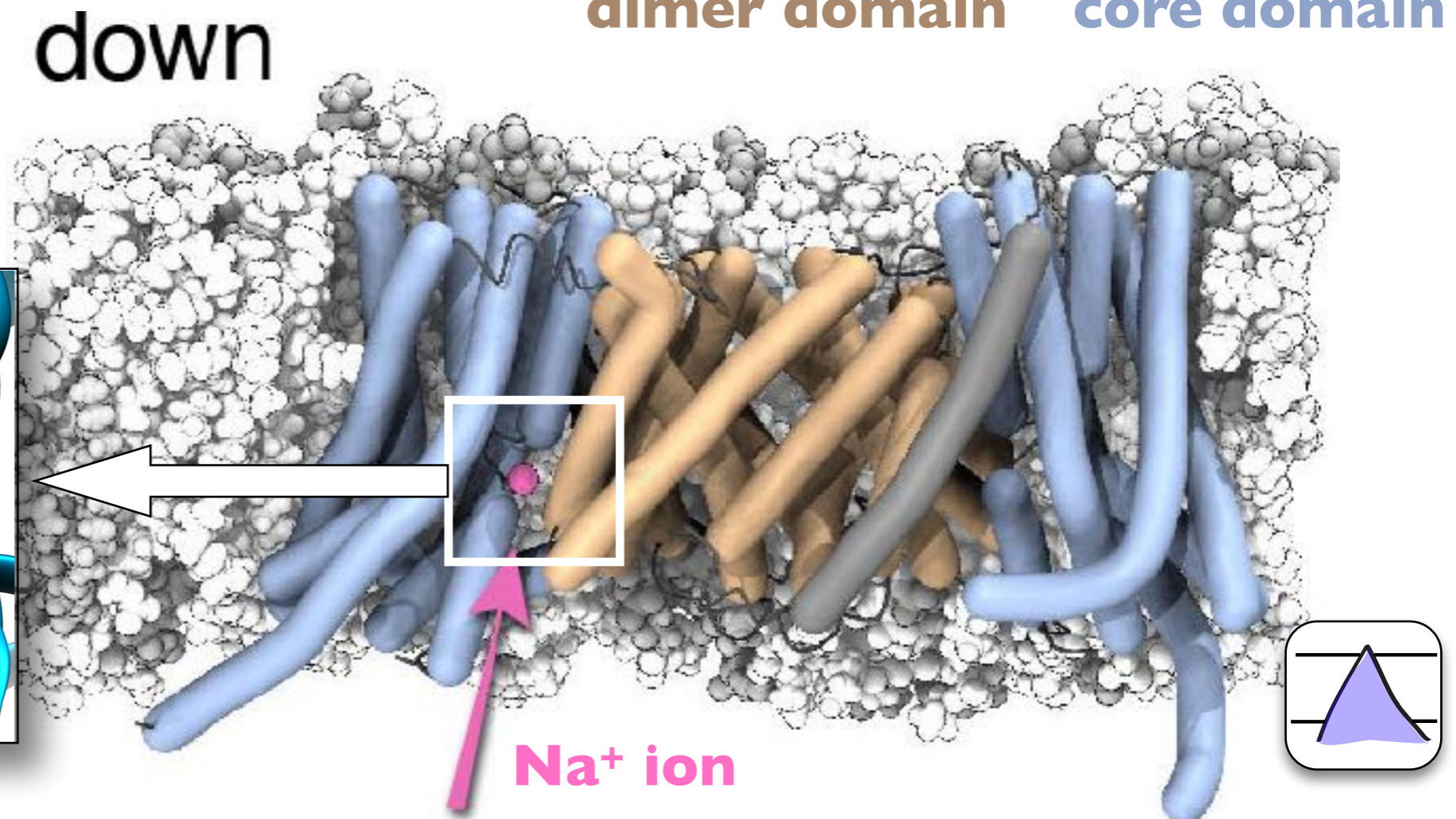
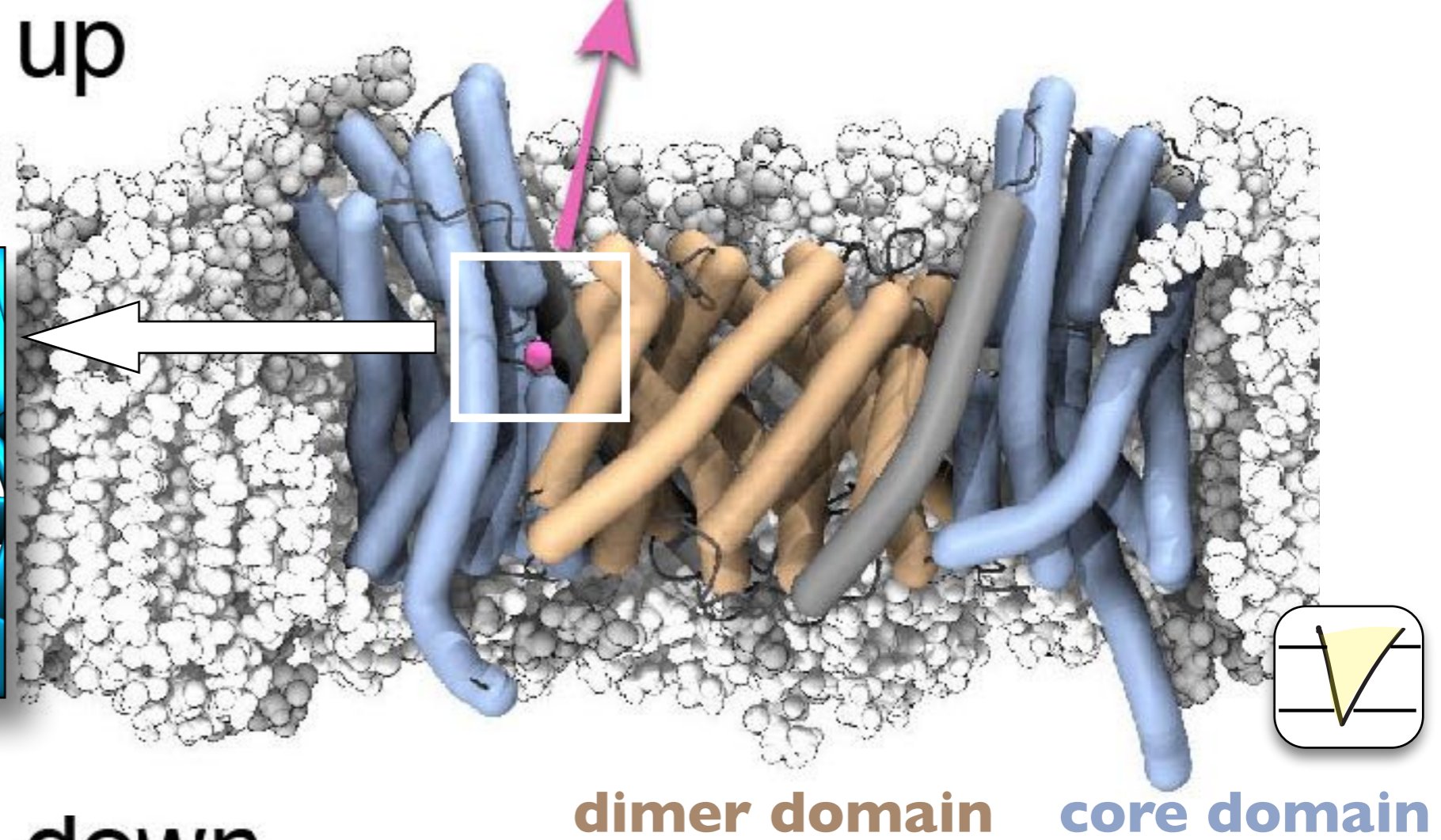


*Nature Struct. Mol. Biol.*, 23 (2016):248–255

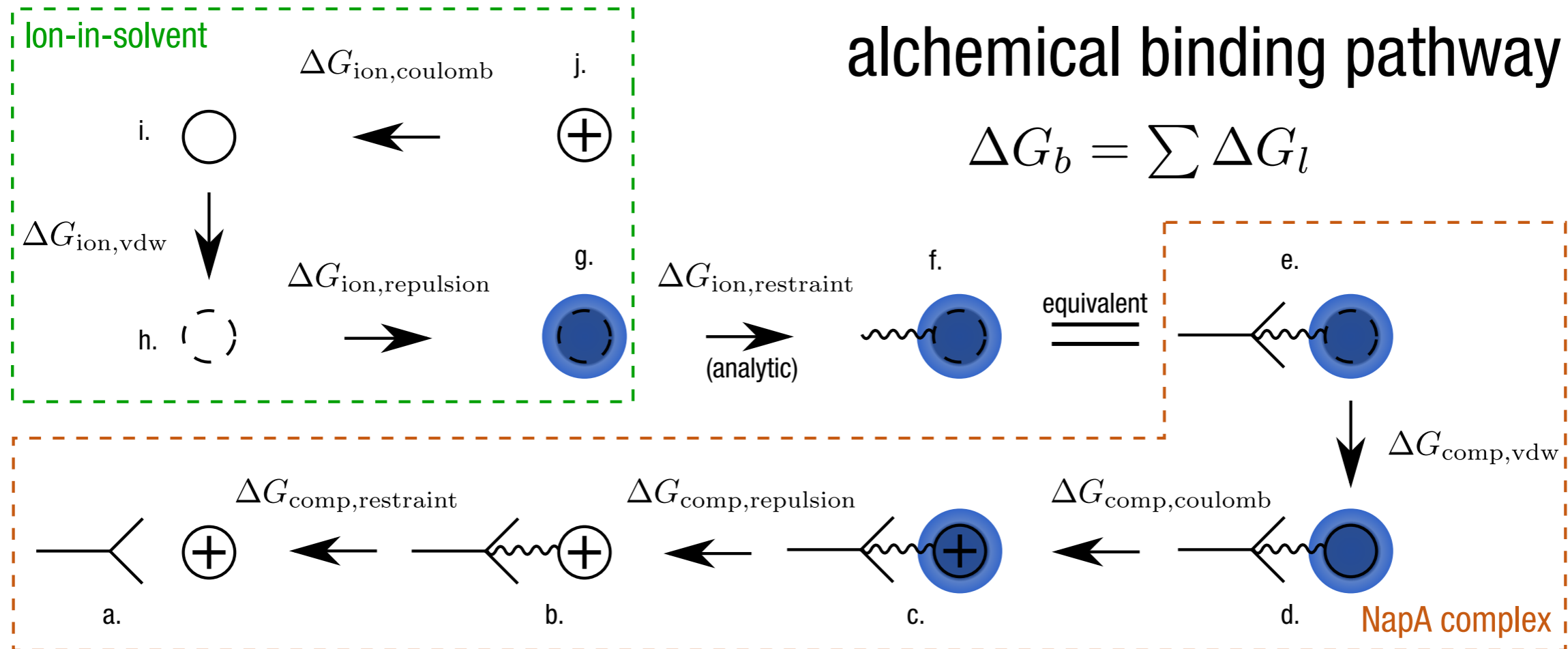
ion binding in different states?



*Nature*, 501 (2013):573–577.



# Absolute binding free energies: alchemistry



- Windowed alchemical free energy calculations (TI or MBAR)
- 150 ns – 250 ns *per lambda window* (Coulomb/vdW decoupling) for 21+21 windows... ~8  $\mu\text{s}$  (!)
- Position restraints (and analytical removal)
- Additional repulsive ion-ion potential to enforce one-ion occupancy (rigorously removed in calculation)

# Amount of raw free energy data

NapA FEP simulations (one free energy)

	windows	time $\mu$ s	size GB	total time $\mu$ s	total size (GB)
<b>VDW</b>	21	0.25	3.865	5.25	81.165
<b>Coulomb</b>	21	0.25	3.865	5.25	81.165
<b>repulsion</b>	3	0.01	0.16	0.03	0.48
<b>restraint</b>	11	0.01	0.16	0.11	1.76
				10.64	<b>164.57</b>

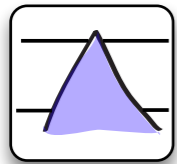
- conformations: IF and OF (2)
- protonation states: 3
- repeats: x2 (some)
- ~12 sets of simulations: ~2 TB (in ~130,000 files)

# Absolute binding free energies: alchemistry

<https://github.com/alchemistry/alchemlyb>

$$\Delta G_i^0 = \Delta G_{\text{protein+ion},i}^0 - \Delta G_{\text{hydration}}^0$$

NapA IF

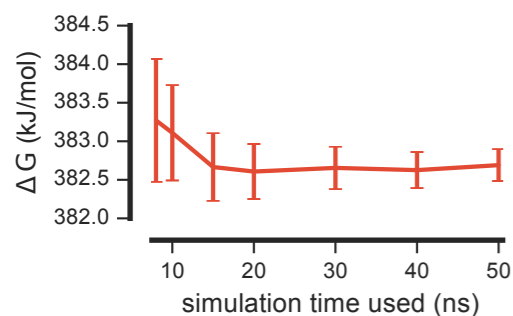


D156	D157	K305	$\Delta G^0$ (kJ/mol)
∅	∅	∅	-103±1
∅	∅	H	-44.6±0.9
∅	H	H	-24.2±13.0

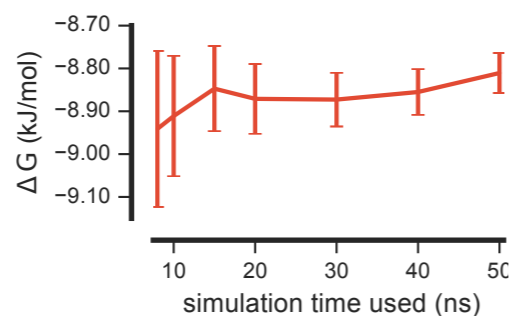
(not binding – ignore state)

## Convergence of ion hydration calculation

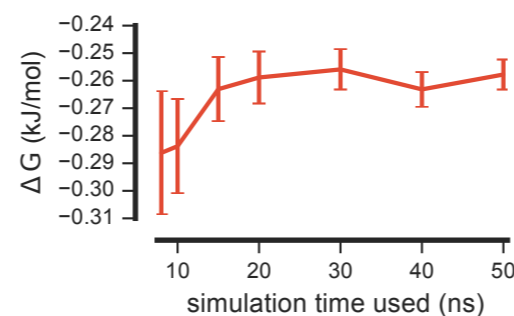
coulomb



vdw



ion-ion repulsion



$\Delta G_{\text{hydration}}$  (kJ/mol)

Coulomb

-382.7±0.2

VdW

8.81±0.05

repulsion

-0.258±0.005

restraint

-17.84

total

-392.0±0.2

Na<sup>+</sup> ion in CHARMM TIP3P water

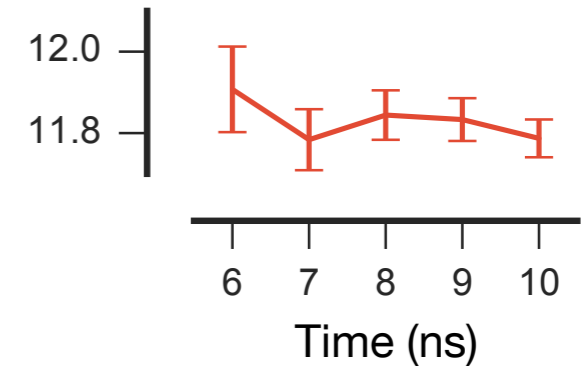
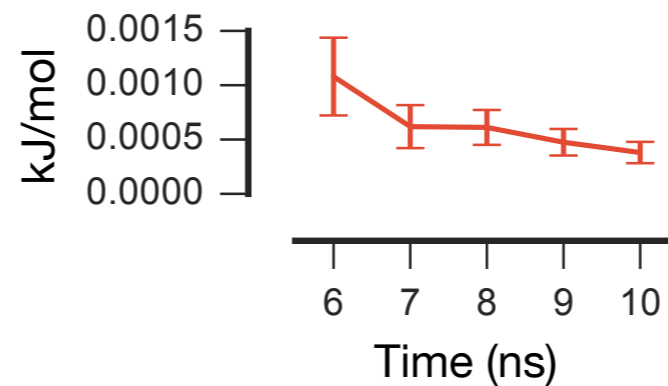
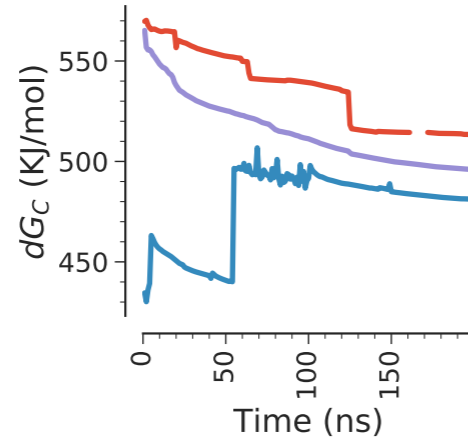
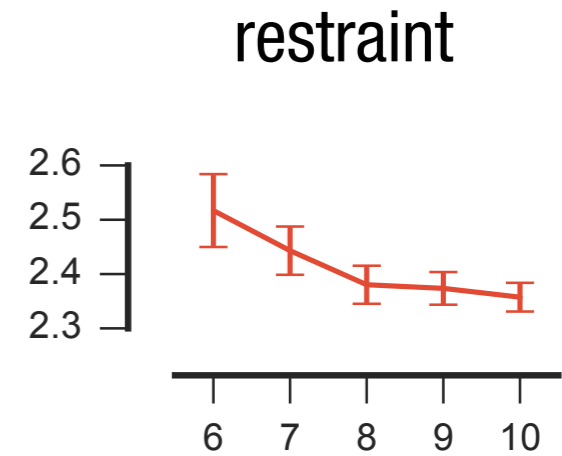
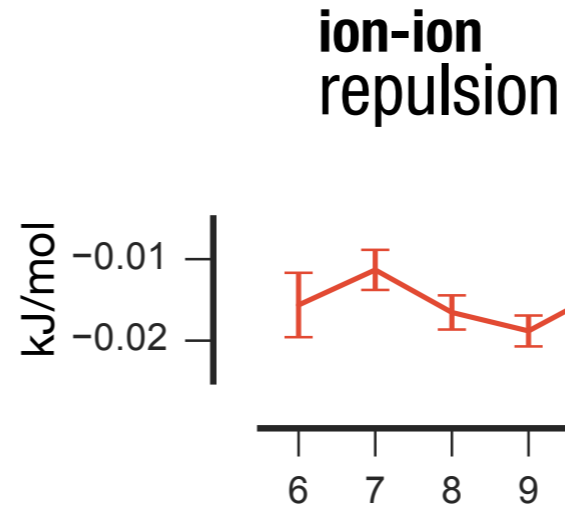
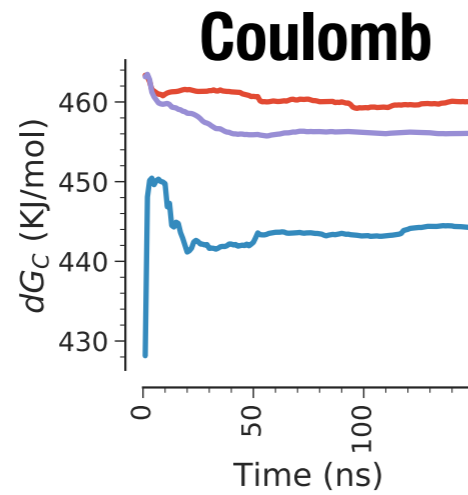
# Convergence of protein-ion calculations

D156 D157 K305

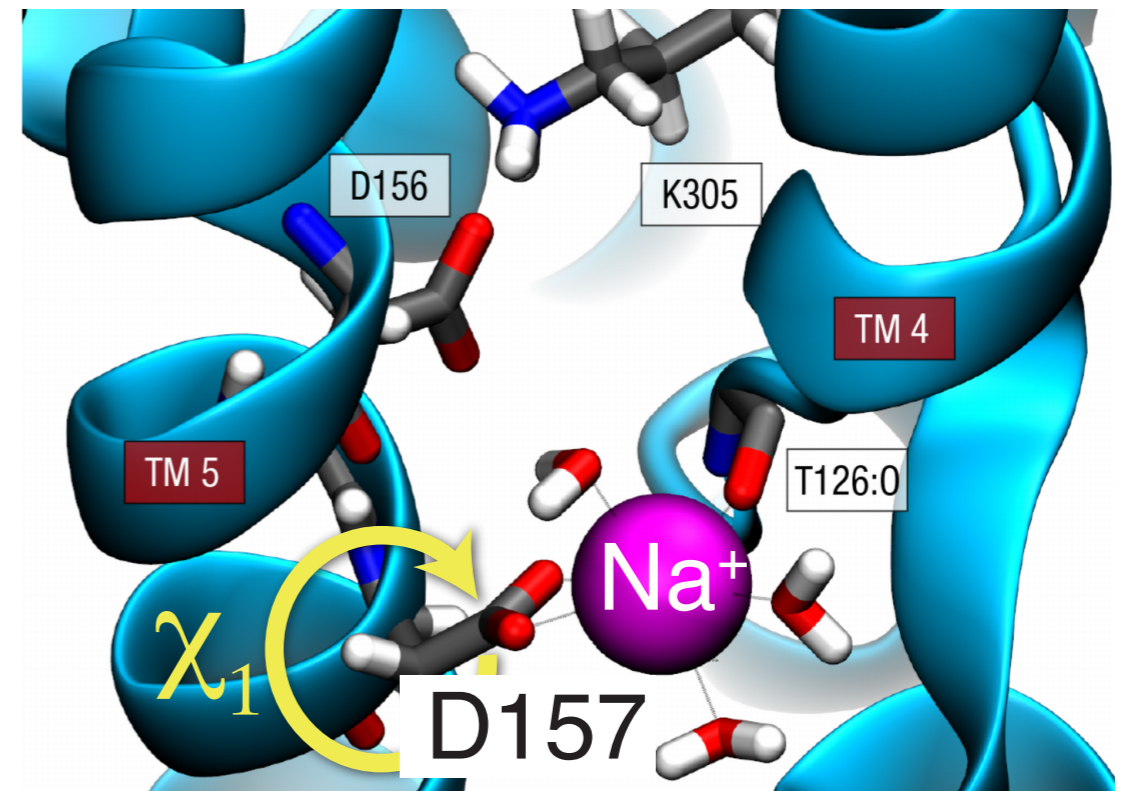
∅ ∅ H

∅ ∅ ∅

—  $\chi_{-180^\circ}$   
—  $\chi_{-90^\circ}$   
— all



- Very slow convergence of Coulomb
- Slow degrees of freedom (e.g., D157  $\chi_1$  dihedral)





# Challenge

- data analysis is cumbersome and slow(ish), even with dask on a 6 core workstation (hours)
- on-demand analysis with all current data/while new data is coming in?
- run on HPC system (e.g. XSEDE PSC Bridges)?